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(54) Method for preparing syndiotactic poly (α -olefin).

(57) A method for preparing a syndiotactic poly(α -olefin) which comprises the step of polymerizing α -olefin in the presence of a catalyst consisting substantially of a transition metal compound having asymmetric ligands, an aluminoxane and a trialkylaluminum having an alkyl group of 2 to 12 carbon atoms, a molar ratio of the transition metal compound having the asymmetric ligands to the aluminoxane being from 1:10 to 1:500, a molar ratio of the transition metal compound having the asymmetric ligands to the trialkylaluminum being from 1:500 to 1:5000.

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BACKGROUND OF THE INVENTION

(a) Field of the Invention

The present invention relates to a method for preparing a poly(α -olefin) having a syndiotactic structure. More specifically, it relates to a method for preparing a poly(α -olefin) by the use of a catalytic system in which a relatively small amount of an expensive aluminosilane is used.

(b) Description of the Related Art

A syndiotactic polypropylene has been known for a long period of time. Such a polypropylene is prepared by polymerization at a low temperature in the presence of a conventional catalyst comprising a vanadium compound, an ether and an organic aluminum compound, but the polypropylene prepared by this process is poor in syndiotacticity and therefore it is scarcely considered that the thus prepared polypropylene exerts its inherent syndiotactic characteristics.

On the contrary, a polypropylene having a good tacticity, i.e., a syndiotactic pentad fraction of more than 0.7 has been discovered for the first time by J. A. Ewen et al. which can be obtained by the use of a catalyst comprising a transition metal compound having asymmetric ligands and an aluminosilane [J. Am. Chem. Soc., 110, 6255-6256 (1988)].

The present inventors have found that a poly(α -olefin) having a syndiotactic structure can be likewise obtained by polymerizing another α -olefin by the similar process.

The above-mentioned J. A. Ewen et al. method is excellent, because the activity of the catalyst per unit weight of the transition metal is good and the tacticity of the obtained polymer is high. However, in this method, it is required that the aluminosilane is usually used in a ratio of 1000 mols or more per mol of the transition metal, and thus it cannot be considered that the activity of the catalyst per unit weight of the aluminosilane is sufficiently high. In addition, since the aluminosilane is expensive, the obtained syndiotactic polypropylene is also expensive. The aluminosilane is difficult to be removed from the polymer, as compared with a usual alkyl aluminum, and in case that it is necessary to decrease the residue of the catalyst, the method used for removing the alkyl aluminum is scarcely applicable.

On the contrary, a method has been suggested in which a polymer having good granular properties is obtained in a high yield by (co)polymerizing an α -olefin, particularly ethylene or ethylene and another α -olefin by the use of a highly active catalyst comprising a specific organic zirconium compound

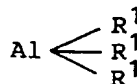
and a reaction mixture which comprises a reaction mixture of an aluminosilane and a trialkylaluminum and an inorganic compound having a surface hydroxyl group (Japanese Patent Laid-open No. 61-276805). However, in the above-mentioned catalyst, a molar ratio of trialkylaluminum to the aluminosilane is 1 or less, particularly from 0.1 to 0.5, and the amount of the aluminosilane to the transition metal compound is not decreased.

Furthermore, a (co)polymerization process of an α -olefin has been suggested in which there is used a catalyst comprising a transition metal compound in the group IVb of the periodic table, 3 milligram atom/liter or less of an aluminosilane in terms of an aluminum atom in a reaction system and an organic aluminum compound having a hydrocarbon group other than an n-alkyl group (Japanese Patent Laid-open No. 63-178108). However, this catalyst is only suitable for the preparation of a polymer not having steric regularity such as a (co)polymer of ethylene or ethylene and α -olefin, and it is necessary that the organic aluminum has the hydrocarbon group other than the n-alkyl group.

SUMMARY OF THE INVENTION

An object of the present invention is to provide a method for preparing a poly(α -olefin) having a high syndiotacticity with a high productivity by the use of a catalyst containing a relatively small amount of an aluminosilane.

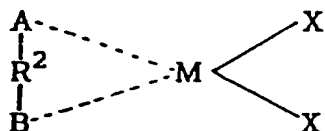
A method for preparing a syndiotactic poly(α -olefin) of the present invention comprises polymerizing an α -olefin in the presence of a catalyst consisting substantially of a transition metal compound having asymmetric ligands, an aluminosilane and a trialkylaluminum represented by the formula



(wherein R^1 is an alkyl group having 2 to 12 carbon atoms), a molar ratio of the transition metal compound having the asymmetric ligands to the aluminosilane being from 1:10 to 1:500, a molar ratio of the transition metal compound to the trialkylaluminum being from 1:500 to 1:5000.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

A transition metal compound having asymmetric ligands which is used in the present invention is preferably represented by the formula



(wherein A and B are mutually different cyclic unsaturated hydrocarbon residues, R² is a hydrocarbon residue having 1 to 20 carbon atoms or a group containing silicon or germanium which connects A with B, X is a halogen atom or a hydrocarbon residue having 1 to 20 carbon atoms, and M is a metal atom selected from the group consisting of titanium, zirconium and hafnium).

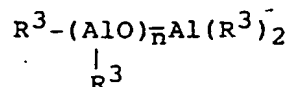
Examples of A and B include monocyclic and polycyclic unsaturated hydrocarbon groups having 5 to 30 carbon atoms, and typical examples thereof include a cyclopentadienyl group, cyclopentadienyl groups in which a part or all of hydrogen atoms are substituted by alkyl groups having 1 to 10 carbon atoms (the terminal of this alkyl group may also be bonded to the cyclopentadiene ring), polycyclic aromatic hydrocarbon groups such as an indenyl group and a fluorenyl group, and polycyclic aromatic hydrocarbon groups in which a part or all of hydrogen atoms are substituted by alkyl groups having 1 to 10 carbon atoms.

Preferable examples of R² include a dialkylmethylene group, a dialkylsilylene group and a dialkylgermylene group, and for example, there can be used compounds represented by (R⁴)₂C<, (R⁴)₂Si< and (R⁴)₂Ge< (wherein R⁴ is hydrogen or an alkyl group having 1 to 20 carbon atoms, and the two R⁴s may be identical or different and may be mutually bonded), and an ethylene group represented by -C(R⁴)₂-C(R⁴)₂- (wherein R⁴ is as defined above). Examples of X include fluorine, chlorine, bromine, iodine, alkyl groups such as methyl, ethyl, propyl and butyl, and an aromatic compound such as a cyclopentadienyl group. In particular, chlorine and the methyl group are preferable.

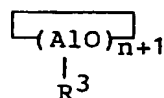
Typical examples of the above-mentioned transition metal compound include isopropylidene(cyclopentadienylfluorenyl)zirconium dichloride and isopropylidene(cyclopentadienylfluorenyl)hafnium dichloride mentioned in the above J. A. Ewen et al. literature, and methylphenylene(cyclopentadienylfluorenyl)zirconium dichloride, methylphenylmethylene(cyclopentadienylfluorenyl)hafnium dichloride, diphenylmethylene(cyclopentadienylfluorenyl)zirconium dichloride and diphenylmethylene(cyclopentadienylfluorenyl)hafnium dichloride mentioned in EP 387690 and EP 387691.

Examples of the aluminoxane include com-

pounds represented by the formula

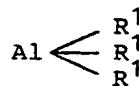


or



(wherein R³ is a hydrocarbon residue having 1 to 3 carbon atoms, and n is an integer of from 1 to 50). The aluminoxane in which R³ is a methyl group and n is an integer of 5 or more is particularly preferable.

The usable trialkylaluminum is represented by the formula



(wherein R¹ is an alkyl group having 2 to 12 carbon atoms), and it may be mixed with another alkylaluminum. Typical examples of the trialkylaluminum include compounds of the above formula in which R¹ is ethyl, propyl, butyl, pentyl, hexyl, heptyl or octyl. This kind of alkyl group is preferably a straight-chain group such as n-propyl, n-butyl, n-pentyl, n-hexyl, n-heptyl or n-octyl. In the present invention, trimethylaluminum is not preferable.

The amount of the aluminoxane to be used is from 10 to 500 mols, preferably 50 to 500 mols per mol of the transition metal compound having asymmetric ligands. When the amount of the aluminoxane is more than 500 mols per mol of the transition metal compound, the decrease of the expensive aluminoxane is insufficient, and when it is less than 10 mols, the activity of the catalyst is unpreferably low. Most preferably, the amount of the aluminoxane is from 100 to 400 mols per mol of the transition metal compound.

The amount of the alkylaluminum to be used is preferably from 500 to 5000 mols per mol of the transition metal compound. When the amount of the alkylaluminum is less than 500 mols, the effect of activity improvement is poor, and when it is more than 5000 mols, any further effect cannot be expected any more and the activity deteriorates

unexpectedly in certain cases. Most preferably, the amount of the alkylaluminum is from 1000 to 4000 mols per mol of the transition metal compound.

In the present invention, the α -olefin having 3 to 25 carbon atoms is preferable, and typical examples thereof include straight-chain α -olefins such as propylene, butene-1, pentene-1, hexene-1, heptene-1, octene-1, nonene-1, decene-1, undecene-1, dodecene-1, tridecene-1, tetradecene-1, pentadecene-1, hexadecene-1 and octadecene-1, and branched α -olefins such as 3-methylbutene-1, 4-methylpentene-1 and 4,4-dimethylpentene-1. These α -olefins can be used singly or in a mixture thereof, and they may be mixed with a small amount of ethylene, when used.

No particular restriction is put on polymerization conditions, and various polymerization techniques are utilizable such as solvent polymerization using an inert medium, bulk polymerization in which there is substantially no inert medium, and gas phase polymerization. Preferably, the polymerization temperature is from -100 to 200°C and the polymerisation pressure is from atmospheric pressure to 100 kg/cm^2 (gauge). More preferably, the temperature is from -100 to 100°C and the pressure is from atmospheric pressure to 50 kg/cm^2 (gauge).

In the thus obtained poly(α -olefin), an intrinsic viscosity measured in a tetralin solution at 135°C is usually from 0.05 to 10 dl/g, and a ratio of the weight average molecular weight to the number average molecular weight of the polymer is usually from 1.5 to 15. The intrinsic viscosity and the molecular weight distribution can be adjusted to desirable values in compliance with a purpose.

The syndiotactic poly(α -olefin) obtained in accordance with the method of the present invention is excellent in balance between stiffness and impact resistance and can provide molded products having excellent transparency. Thus, this kind of syndiotactic poly(α -olefin) is different from an isotactic poly(α -olefin) in characteristics.

In the present invention, it is effective that polymerization is carried out at a low temperature of 100°C or less by the use of a catalyst containing a transition metal compound purified to a high purity (90% or more) in order to further improve the syndiotacticity of the obtained polymer, and it is also effective that the obtained polymer is washed with a hydrocarbon solvent.

Examples of the hydrocarbon solvent include hydrocarbon compounds having 3 to 20 carbon atoms, for example, saturated hydrocarbon compounds such as propylene, propane, butane, pentane, hexane, heptane, octane and nonane, aromatic hydrocarbon compounds such as benzene, toluene, xylene and ethylbenzene, and these compounds in which a part of all of hydrogen atoms

are substituted by fluorine, chlorine, bromine or iodine. Examples of other usable solvents include alcohols having 1 to 20 carbon atoms and ethers and esters having 2 to 20 carbon atoms which can dissolve or disperse a low-molecular atactic component therein. No particular restriction is put on a washing manner, and the washing is usually carried out at a temperature of from 0 to 100°C .

The present invention will be described in detail in reference to examples and comparative examples. However, these examples intend to elucidate the present invention, and they should not be construed to be restrictive.

Example 1

1 mg of isopropyl(cyclopentadienyl-1-fluorenyl)zirconium dichloride, 0.027 g [200 mols per mol of isopropyl(cyclopentadienyl-1-fluorenyl)zirconium dichloride] of methylaluminoxane (polymerization degree 16.2) made by Toso Akzo Co., Ltd and 0.67 g [2000 mols per mol of (cyclopentadienyl-1-fluorenyl)zirconium dichloride] of triethylaluminum were dissolved in 50 ml of toluene, and the mixture was then placed in an autoclave. This isopropyl(cyclopentadienyl-1-fluorenyl)zirconium dichloride was obtained by converting, into a lithium salt, isopropylcyclopentadienyl-1-fluorene synthesized in an ordinary manner, and then reacting the same with zirconium tetrachloride. Afterward, 1.5 kg of propylene was put in the autoclave, and temperature was elevated and polymerization was then carried out at 50°C for 2 hours.

After the polymerization, unreacted propylene was purged, and the resultant powder was taken out, dried and then weighed, whereby 210 g of a polypropylene was obtained.

According to ^{13}C -NMR, the syndiotactic pentad fraction of this polypropylene was 0.84, and an intrinsic viscosity (hereinafter referred to as " η ") measured in a tetralin solution at 135°C was 0.86. Furthermore, a ratio of the weight average molecular weight to the number average molecular weight (hereinafter referred to as " M_w/M_n ") of the polymer measured in 1,2,4-trichlorobenzene was 2.3.

Comparative Example 1

The same procedure as in Example 1 was effected except that no triethylaluminum was used in polymerization, and as a result, 5.1 g of a syndiotactic polypropylene was merely obtained. Furthermore, according to ^{13}C -NMR, the syndiotactic pentad fraction of this polypropylene was 0.62, its η was 0.94, and its M_w/M_n was 2.7.

Comparative Example 2

The same procedure as in Example 1 was effected except that polymerization was carried out using 0.27 g of methylaluminoxane without triethylaluminum, and as a result, 315 g of a polypropylene was obtained.

In the thus obtained polypropylene, a syndiotactic pentad fraction of was 0.85, η was 1.09, and Mw/Mn was 2.1.

Example 2

The same procedure as in Example 1 was effected except that triethylaluminum was replaced with triisobutylaluminum, thereby obtaining 285 g of a polypropylene. In the thus obtained polypropylene, a syndiotactic pentad fraction was 0.84, η was 1.19, and Mw/Mn was 2.2.

Example 3

The same procedure as in Example 2 was effected except that triisobutylaluminum was replaced with tri-n-butylaluminum, thereby obtaining 362 g of a polypropylene. In this polypropylene, a syndiotactic pentad fraction was 0.86, η was 1.15, and Mw/Mn was 2.1.

Comparative Example 3

The same procedure as in Example 2 was effected except that triisobutylaluminum was replaced with trimethylaluminum, but in this case, a polymer was scarcely obtained.

Example 4

The same procedure as in Example 1 was effected except that amounts of triethylaluminum were 550 mols, 1000 mols, 3000 mols, 4000 mols and 5000 mols, and polypropylenes were obtained in amounts of 10 g (syndiotactic pentad fraction was 0.82, and η was 1.08), 15 g (syndiotactic pentad fraction was 0.81, and η was 1.12), 205 g (syndiotactic pentad fraction was 0.85, and η was 1.02), 195 g (syndiotactic pentad fraction was 0.83, and η was 0.92), and 200 g (syndiotactic pentad fraction was 0.84, and η was 0.88), respectively.

Comparative Example 4

The same procedure as in Example 3 was effected except that the amount of triethylaluminum was 450 mols, thereby obtaining 8 g of a polypropylene in which a syndiotactic pentad fraction was 0.65 and η was 0.92.

Example 5

The same procedure as in Example 1 was effected except that amounts of methylaluminoxane were 10 mols, 50 mols, 100 mols, 400 mols and 450 mols, so that polymers were obtained in amounts of 11 g, 30 g, 150 g, 230 g and 215 g, respectively. These polymers had syndiotactic pentad fractions of 0.80, 0.80, 0.81, 0.83 and 0.84, respectively as well as η s of 1.02, 1.08, 1.01, 0.93 and 0.89, respectively.

Comparative Example 5

The same procedure as in Example 4 was effected except that the amount of methylaluminoxane was 5 mols, but in this case, a polymer was scarcely obtained.

Example 6

The same procedure as in Example 1 was effected except that propylene was replaced with 4-methylpentene-1 and octene-1, obtaining 1.2 g of poly(4-methylpentene-1) and 2.8 g of poly(octene-1). These polymers had η s of 0.21 and 0.23, respectively. According to ^{13}C -NMR, it was confirmed that each of these polymers had a syndiotactic structure.

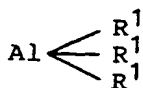
Example 7

The same procedure as in Example 1 was effected except that in place of propylene, propylene and butene-1 were copolymerized in a weight ratio of 85:15, thereby obtaining 115 g of a copolymer. In this copolymer, the content of propylene units was 91% by weight and η was 1.05. Furthermore, a ratio of a peak intensity observed at 20.2 ppm in accordance with ^{13}C -NMR to the total peak intensity attributed to the methyl groups of propylene was 0.68.

According to the method of the present invention, a poly(α -olefin) having a high tacticity can be prepared in a high yield per unit weight of a transition metal of a catalyst system by the use of a catalyst which permits decreasing the amount of an expensive aluminoxane to be used.

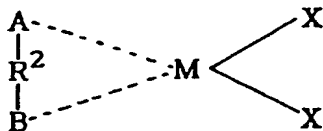
Claims

1. A method for preparing (co)polymer of an α -olefin having a syndiotactic structure which comprises the step of polymerizing said α -olefin in the presence of a catalyst consisting substantially of a transition metal compound having asymmetric ligands, an aluminoxane and an alkylaluminum represented by the formula



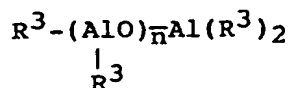
(wherein R¹ is an alkyl group having 2 to 12 carbon atoms), a molar ratio of said transition metal compound having the asymmetric ligands to said aluminosilane being from 1:10 to 1:500, a molar ratio of said transition metal compound having the asymmetric ligands to said alkylaluminum being from 1:500 to 1:5000.

2. The method according to Claim 1 wherein said molar ratio of said transition metal compound having the asymmetric ligands to said aluminosilane is from 1:50 to 1:500.
3. The method according to Claim 1 wherein said molar ratio of said transition metal compound having the asymmetric ligands to said aluminosilane is from 1:100 to 1:400.
4. The method according to Claim 1 wherein said transition metal compound having the asymmetric ligands is a compound represented by the formula

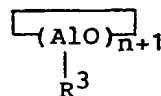


(wherein A and B are mutually different cyclic unsaturated hydrocarbon residues, R² is a hydrocarbon residue having 1 to 20 carbon atoms or a group containing silicon or germanium which connects A with B, X is a halogen atom or a hydrocarbon residue having 1 to 20 carbon atoms, and M is a metal atom selected from the group consisting of titanium, zirconium and hafnium).

5. The method according to Claim 1 wherein said aluminosilane is a compound represented by the formula



or



(wherein R³ is a hydrocarbon residue having 1 to 3 carbon atoms, and n is an integer of from 1 to 50).

6. The method according to Claim 1 wherein the alkyl group of said trialkylaluminum is a straight-chain or a branched alkyl group.
7. The method according to Claim 1 wherein said α-olefin is one of straight-chain and branched α-olefins having 3 to 25 carbon atoms, a mixture thereof, or a mixture of them and ethylene.

(19)



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EUROPEAN SEARCH REPORT

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DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
Y	EP-A-0 287 666 (MITSUI PETROCHEMICAL INDUSTRIES LTD.) * claim 1 *	1-7	C08F10/00 C08F4/602
D	& JP-A-63 178 108 (...) ---		
Y	JOURNAL OF THE AMERICAN CHEMICAL SOCIETY, vol. 110, 1988, GASTON, PA US pages 6255 - 6256; J.A. EWEN: 'SYNDIOSPECIFIC PROPYLENE POLYMERIZATION WITH GROUP 4 METALLOCENES.' * the whole document *	1-7	

			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
			C08F
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 27 AUGUST 1992	Examiner SERRAVALLE M.
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document			